### **Dissolution Method Development Combination Product**

# Case 1 Study

Justification of an Alternative Drug Release Method Using Taguchi Method for API-FDC Tablets



### General Flow for Analytical Method Development

- A. Identify potential method variables
- B. Perform preliminary risk assessment for method
- C. Classification of method variables (X, C, N)
- D. Identify preliminary controls for experiment(s) including factors which will be set as constants
  - Note: need to understand how & why
- E. Conduct screening experiment(s) to identify important parameters
  - Note: need to understand how & why
  - One Factor at a Time Experiments (OFAT) or screening depending on variable
- F. Full-scale optimization of method through DOE
- G. Develop control strategy for method

#### Identifying Potential Method Variables Example (Ishikawa Diagram)



#### **Define the Design Space**

#### State Which Parameters Were Investigated & Why

oWhich parameters held constant oWhich parameters were varied oWhich parameters are critical

## Failure Mode Effects Analysis (FMEA)

### A-priori

FMEA: Break down large complex processes into manageable steps

- Evaluation of potential failure modes for drug release method (holistically but with emphasis in APIFDC Tablets)
- The likely effect on CQAs
- Once failure modes are established, risk reduction can be used to eliminate, reduce or control the potential failures
- □ FMEA relies on process understanding
- Summarize the important modes of failure, the likely effects of these failures, factors causing these failures, and our ability to detect these failures

## Preliminary Risk Assessment

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ace primerate and the second s	1. Administration	2. List "Item(s)" 3.	List "Ite	em Funcl	tion(s)" 4. Develop DFMEA	5. Man	age Recommendations	🛃 Design Analysis 📢	ු Se	ttings 🔷	Data Check	
dissolution.fmea	Design Subsystem: Design Compo								onen	t:		_
ve Libraries	Item: 1. Dissol	ution Equipment										
Aerospace Design FMEA Libra Aerospace Process FMEA Libra Automotive Design Library Automotive Process Library	Item Function: 1.											_
	Potential Failure Modes	Potential Effect(s) of Failure	Sev	Class	Potential Cause / Mechanism of Failure	Occur	Current Design Controls (Prevention)	Current Design Controls (Detection)	Detec	R.P.N.	Recommended Action(s)	
) efense Design Library ) efense Process Library  estronic EMEA Library	Basket Mesh Size Wrong	Inadequate release	3		QC of baskets; regional requirements	1	Inspection of baskets upon receipt	Inspection of baskets upon use for USP/JP/EP requirements	2	6	12. Specify basket specs to follow	
eneral Design Library eneral FMEA Library	Standards Preparation ₩rong	Incorrect results	3		Various	1	Verification of standards prep after use	Controlled weight of standards on sample	1	3		
ieneral IT Library ieneral Process Library							Verification of calculations	prep plan				
lachinery Automotive Library	Method to Lower Baskets Inappropriate	∀ariable results	2		Lack of experience	1	Fixed procedure for lowering baskets	None	2	4	13. Review procedure for lowering baskets	
	Baskets Connection Inadequate	∀ariable resutts	2		Slipping of shaft	2 Equipment maintenance every 6? months beginning and end of run		2	8			
	Bubbles on Baskets	∀ariable resutts	3		Degassing inadequate	1 Degas media before use		Visual check for bubbles	1	3		
	Dissolution baskets vendor	∨ariable resutts	1		Lab controls inadequate	1	Label vessels per original vendor and specify in method notes	Check during test and document on test sheet	1	1	14. DOE: Various vessels	
							Conduct experiment with various makes					
	Temperature of media during test too high or	Variable results	2		Hot/Cold spots in units	1	Calibrated temperature sensors & equipment	Monitor temperature throughout test	1	2		
	100 1097						Verification of temperature at 37C before start					
	Basket/Paddle speed Too High	Release rate too high	3		Wrong set points	1	Specify speed in procedure & document	Control charts placed on equipment to verify	1	3	15. DOE: basket/paddle speed	
							DOE work on effect of speed	speed during test				
					Disso bath connection to sampling unit not well controlled	1	Specify speed in procedure & document	Control charts placed on equipment to verify	2	6	15. DOE: basket/paddle speed	
							Verification of equipment communication to ensure no override of disso bath by autosampler	speed during test				
	Basket/Paddle Speed Too Low	Release rate too low	3		Wrong set points	1	Verify setpoints before start of test & document	Control charts placed on equipment to verify speed during test	1	3	15. DOE: basket/paddle speed	
					Disso bath connection to sampling	1	Verification of equipment	Control charts placed	1	3	15. DOE: basket/paddle	

- Drug release method for API-FDC tablets to identify high-risk steps or critical parameters
- Define, assess and prioritize risks
- Monitor the effectiveness of risk control activities

## Method Development Considerations



# What we know? CQAs?

Know/CQAs	Proposals/Solutions/Investigations
API- more stable in solution at pH 3 – 5; pKa (calculated) = 4.2;	A pH of 6.8 is a good option based on pKa value (~2 pH units above its pKa)
Active is sensitive to moisture	Package with desiccants
IR formulation	Film coated to protect from light
Monitoring Product quality and performance after changes such as manufacturing process, scale-up	Multiple lots of product available? What are the differences in raw materials, processes, etc.? This could facilitate decisions on discriminatory capability?
Guidance for Industry: Dissolution testing of IR Solid Oral Dosage Form	Appendix A: Sink conditions are desirable but not mandatory A pH 6.8 should be employed: a higher pH should be justified on a case-by-case basis
Dissolution	Apparatus 1 (50-100 rpm) or Apparatus 2 (50- 75 rpm) – simple, robust
	Addition of surfactant acceptable for water insoluble or sparingly water soluble. The need and amount should be justified.
Physiological pH range of 1.2 – 6.8, solubility low	Use surfactant: Use minimally 100% Saturated solubility
Specifications	TBD

### Establishment Of The Alternative Condition For API-FDC Tablets

According to the FDA guidance, we investigated the alternative condition in order to characterize the quality of the product

#### <Investigated Conditions>

- pH of media: 6.0 and 6.8
- Surfactant: SDS, Tween80, CTAB
- Apparatus: 1 and 2

#### <Points for selecting condition>

- Lower dissolution rate @ early time points (slower release)
- 85% dissolution rate @ 45 min
- Observation (artifacts such as coning, undissolved materials, large particles floating, etc.)

#### <Approaches>

- OFAT or screening depending on variable
- Statistical methods such as Taguchi

### A Modified 4-Factor Central Composite Design Was Used

	Coded Unit					Actual Cond	X4 = Vendor		
Pattern	X1	X2	X3	X4	X1 (pH)	X2 (Conc.)	X3 (RPM)	X4	
	-1	-1	-1	L1	6.7	0.9	50	L1	
+	-1	-1	1	L2	6.7	0.9	75	L2	
-+-	-1	1	-1	L2	6.7	1.2	50	L2	
-++	-1	1	1	L1	6.7	1.2	75	L1	
+	1	-1	-1	L2	6.9	0.9	50	L2	
++	1	-1	1	L1	6.9	0.9	75	L1	
++	1	1	-1	L1	6.9	1.2	50	L1	
+++	1	1	1	L2	6.9	1.2	75	L2	
a00	-1.2	0	0	L2	6.65	1	62.5	L2	
A00	1.2	0	0	L2	6.95	1	62.5	L2	
0a0	0	-1.2	0	L2	6.8	0.85	62.5	L2	
0A0	0	1.2	0	L2	6.8	1.25	62.5	L2	
00a	0	0	-1.2	L2	6.8	1	45	L2	
00A	0	0	1.2	L2	6.8	1	80	L2	
000	0	0	0	L1	6.8	1	62.5	L1	
000	0	0	0	L2	6.8	1	62.5	L2	

#### Contour Plots: API-FDC Tablets, Drug Release

#### Contour Plot for RPM vs. pH code (15 min)



Contour Plot for RPM vs. concentration code (15 min)



Contour Plot for RPM vs. pH Code (20 min)



Contour Plot for RPM vs. concentration code (20 min)



#### Alternative Condition for API-FDC

Mild test conditions such as Apparatus 2 @ 50 rpm and lower conc., Tween such as 1.0% could be selected as an alternative method

Solubility of API in the various conc. of Tween80 solutions (37°C)



Solution	Solubility (mg/mL)				
Phosphate buffer (pH 6.8)	-				
+Tween80 0%	0.021				
+Tween80 0.5%	0.063				
+Tween80 0.6%	0.072				
+Tween80 0.7%	0.076				
+Tween80 0.8%	0.087				
+Tween80 1.0%	0.104				
+Tween80 1.2%	0.123				
+Tween80 2.0%	0.197				

### Drug Release Methods for APIFDC Tablets Summary

Parameters	Method 1 (Current)	Method 2 (Investigated)	Method 3 (Alternative)		
Apparatus:	USP Dissolution Apparatus 2 (paddles)	USP Dissolution Apparatus 2 (paddles)	USP Dissolution Apparatus 2 (paddles)		
Rotation:	50 RPM ± 4%	50 RPM ± 4%	50 RPM ± 4%		
Temperature:	$37^{\circ}C \pm 0.5^{\circ}C$	$37^{\circ}C \pm 0.5^{\circ}C$	$37^{\circ}C \pm 0.5^{\circ}C$		
Medium:					
Drug Release:	900 mL of pH 6.8, potassium phosphate buffer + 2% SDS	900 mL of pH 7.8, potassium phosphate buffer	900 mL of pH 6.8, potassium phosphate buffer + 1% Tween80		
Rationale       Drug release too rapid         (95% LC in 10 min)       May not provide         discriminatory power       Image: Compare the second se		Visual observations show significant method artifacts such as un- dissolved materials, large particles and film components floating during testing with and/or without surfactant	Slower profiles Visual observations doesn't confirm method artifact Discriminatory power against manufacturing process/changes in vendor excipients		
Filter:	35 micron polyethylene	35 micron polyethylene	35 micron polyethylene		
Analytical Finish:		UPLC			
Rationale – Analytical Finish	Justified based on A	PI, FDC Stability, Compariso	n of UV & UPLC, etc.		



# Case 2 Study

## Robustness Evaluation for Related Substance Method Using DOE



### How Was The Experimental Design Chosen For API-FDC Tablets RS Method?

- Select the number of factors to be explored and the objectives of the experiment
- Collaboration between Analytical Lab & Statisticians
- The Plackett-Burman design was selected based
  - On the number of factors to be evaluated
  - Multifactor-designed (matrix) experiment:
    - more efficient, cost-effective, and informative
  - Very effective screening design when only main effects are of interest-
    - ✤ As it was the case for APIFDC Tablets RS method
  - Robustness/Ruggedness evaluation

# Plackett-Burman Design

<ul> <li>Image: A start of the start of</li></ul>	パターン	Mobilep hase pH	Salt Conc of	Gradien t Time	Mobilep hase A.,	Mobilep haseB	Mobilep hase	Temp.	Wavelen gth	Flow rate	column	Instrum ent
1	++-+++-	-1	-1	1	-1	-1	1	-1	1	1	L2	L1
2	+-++++	-1	-1	1	-1	1	1	1	-1	-1	L1	L2
3	****	1	1	1	1	1	1	1	1	1	L2	L2
4	+-++++	-1	-1	-1	1	-1	-1	1	-1	1	L2	L2
5	-++++	-1	1	1	1	-1	-1	-1	1	-1	L1	L2
б	+-++++	1	-1	1	1	1	-1	-1	-1	1	L1	L1
7	-+-++++-	-1	1	-1	1	1	1	-1	-1	-1	L2	L1
8	++-++	1	-1	-1	-1	1	-1	-1	1	-1	L2	L2
9	+++++-	1	1	1	-1	-1	-1	1	-1	-1	L2	L1
10	+++-+-+	1	1	-1	-1	-1	1	-1	-1	1	L1	L2
11	-++-+++	-1	1	-1	-1	1	-1	1	1	1	L1	L1
12	++-+++	1	-1	-1	1	-1	1	1	1	-1	L1	L1

# Results and Conclusions RS Method

- DOE analysis indicates that for the key resolutions (U-6 vs API and C-1 vs FDC), the only factor that has an effect is instrument.
  - 1. Generally the L2 instrument (Agilent 1200) gave higher resolution than the L1 (Waters UPLC)
  - 2. All the resolutions were good (>3).
- For the other minor peaks the key effect was pH and sometimes temperature although pH effect was not consistent
  - All resolutions were very good
- Based on above, instrument was further investigated through a "validation" study
  - 1. Results confirmed the DOE results
  - 2. For API the key resolution mean(rsd) was 6.27 (.19) for Waters and 3.83 (1.2) for Agilent.
  - 3. For FDC the key resolution was 4.20 (.06) for Waters and 6.25 (.94) for Agilent
  - 4. Waters UPLC seems to have better precision (e.g., for resolution)

#### Was robustness established for this method?

- Instrument L1 absolutely
- L2 gave higher resolution than L1 but acceptable resolution obtained

## Major Peaks – API-FDC Tablets - RS



## Minor Peaks – API-FDC Tablets - RS

